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Cite the difference  
between atomic mass  
and atomic weight.

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4.2 For some  
hypothetical metal, the  
equilibrium number of  
vacancies at 900 C is  
2.3  $10^{25} \text{ m}^{-3}$ . If the  
density and atomic  
weight of this metal  
are 7.40  $\text{g/cm}^3$  and

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85.5 g/mol,  
respectively, calculate  
the fraction of  
vacancies for this  
metal at 900 C.

Solution This problem  
is solved using two  
steps: (1) calculate the  
total number of lattice  
sites in silver,  $N$ , using

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